

Teaching Non-Ideal Reactors with CFD Tools

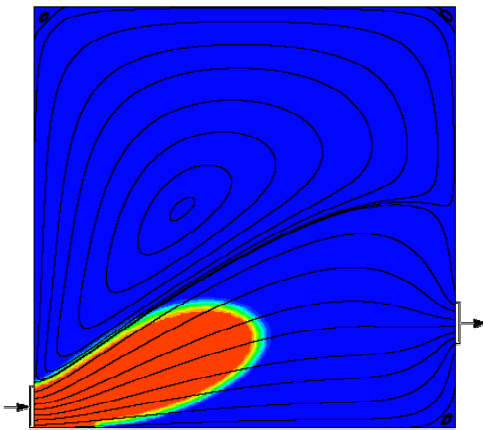
Picture Gallery

The following pictures illustrate the concentration evolution of a tracer inside the reactor. Initially, the reactor is full of **water (blue)**, and a step profile in the concentration of a **tracer (red)** is imposed at the inlet boundary. The reduced time is defined as $\theta = t/\tau$, where τ represents the space-time.

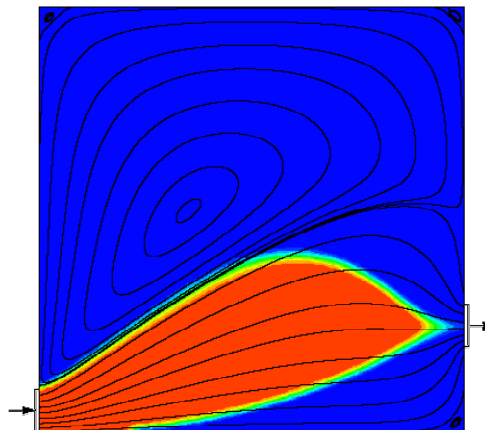
The **streamlines** are shown in black.

$L/H=1$ $Re=10$

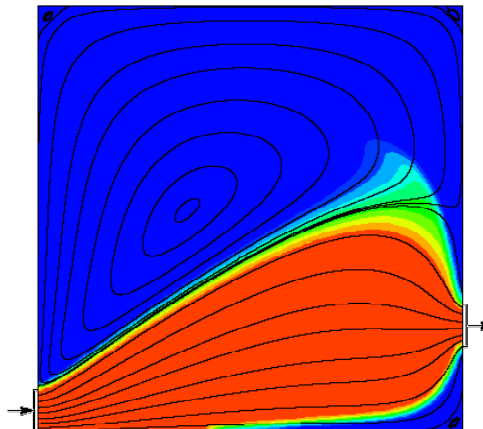
$\theta = 0.1$



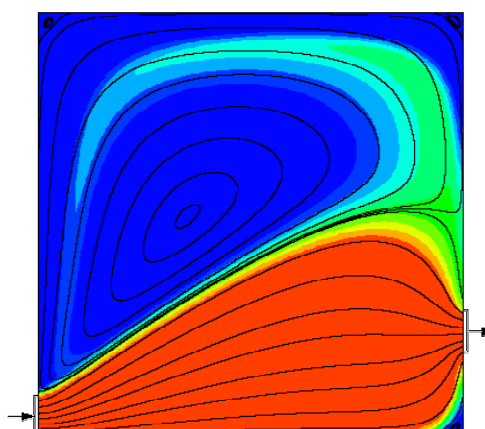
$\theta = 0.22$



$\theta = 1$



$\theta = 5$



These pictures clearly illustrate that approximately at $\theta = 0.22$ one starts to 'see' **tracer** at the reservoir outlet. In addition, even for a very long time of operation (about 5 times the residence time), the reservoir is not completely full of **tracer**, due to the large stagnant zone.

[Go to top](#)

Now, we illustrate the steady-state concentration field of a **reagent** (orange) which undergoes a 1st-order irreversible reaction, originating a **product** species (green). The influence of the Damkohler number, $Da = k\tau$, is illustrated.

$L/H=1$ $Re=10$

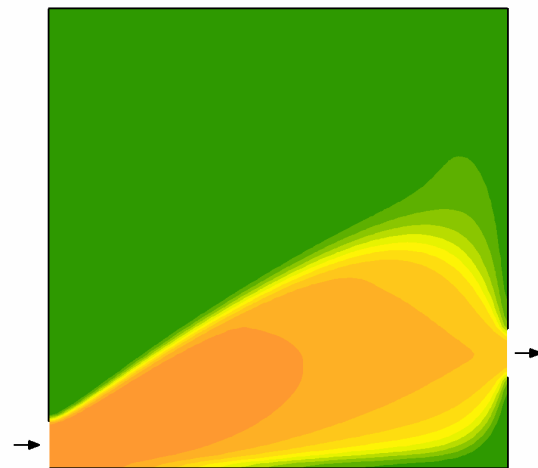
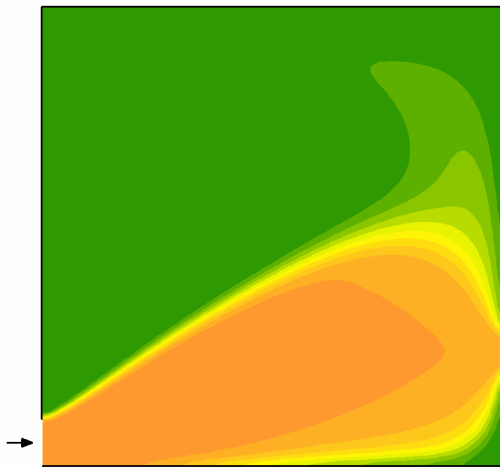
Concentration of reagent



0 0.2 0.4 0.6 0.8 1

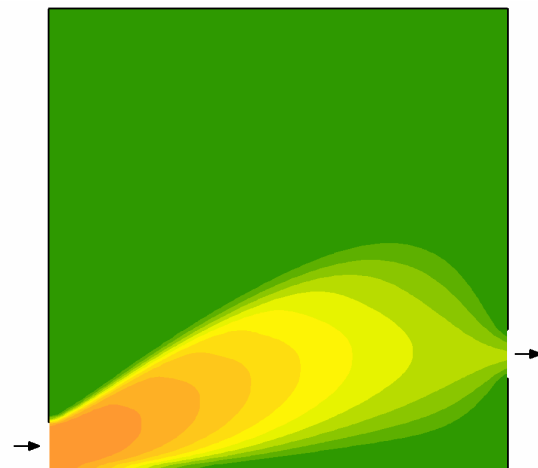
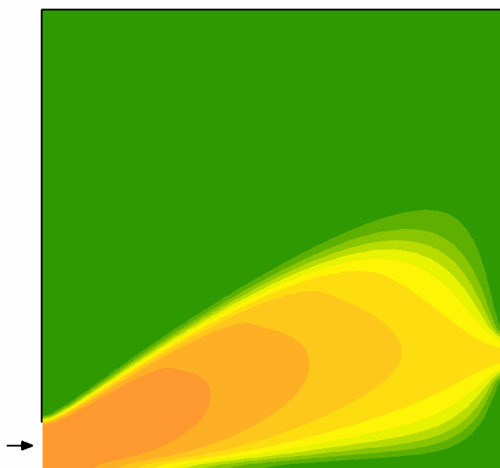
$Da = 0.5$

$Da = 1$



$Da = 2$

$Da = 5$



[Introduction](#)

[Geometry](#)

[Picture Gallery](#)

[Movie Gallery](#)